Chapter 2. Atoms, Molecules and Ions

Dalton's Atomic Theory

Each element is composed of extremely small particles called atoms.

An atom of the element oxygen

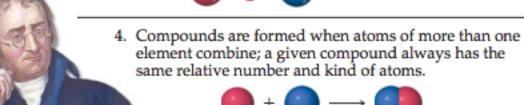
An atom of the element nitrogen

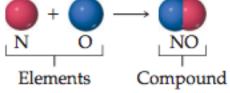
All atoms of a given element are identical, but the atoms of one element are different from the atoms of all other elements.



Atoms of one element cannot be changed into atoms of a different element by chemical reactions; atoms are neither created nor destroyed in chemical reactions.

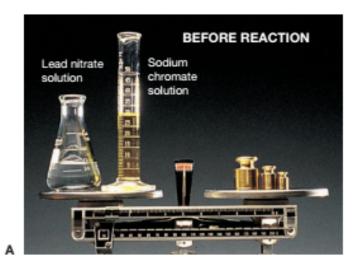
Oxygen





Nitrogen

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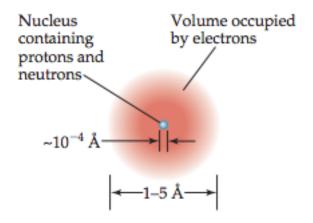


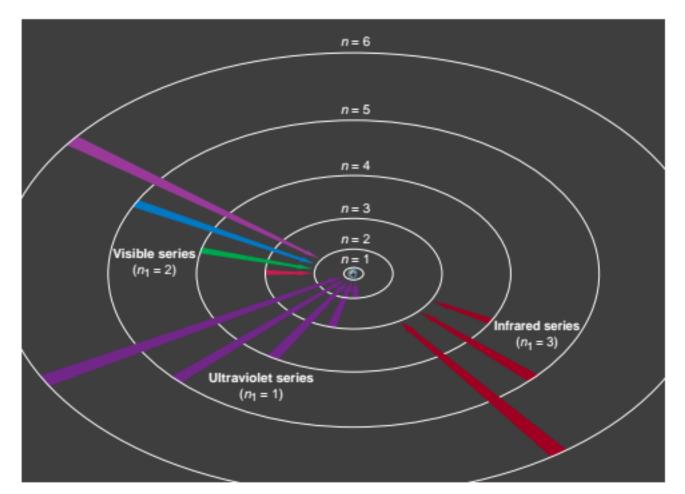
Solid lead chromate in sodium nitrate solution

AFTER REACTION

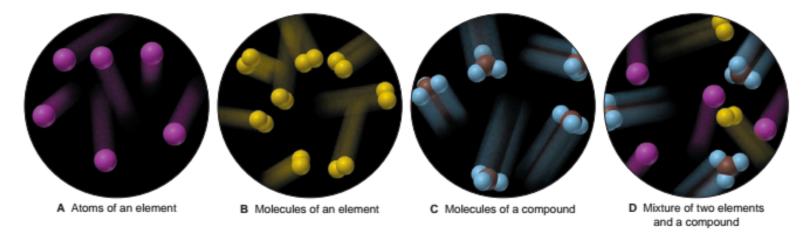
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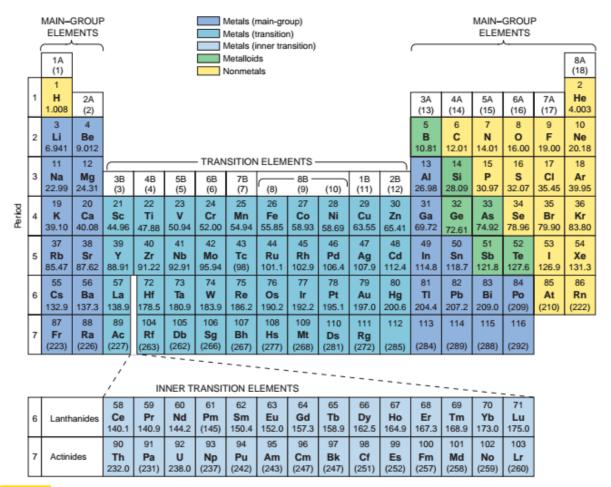


Figure 2.9 The modern periodic table. The table consists of element boxes arranged by increasing atomic number into groups (vertical columns) and periods (horizontal rows). Each box contains the atomic number, atomic symbol, and atomic mass. (A mass in parentheses is the mass number of the most stable isotope of that element.) The periods are numbered 1 to 7. The groups (sometimes called families) have a number-letter designation and a new group number in parentheses. The A groups are the main-group elements; the B groups are the transition elements. Two series of inner transition elements are

placed below the main body of the table but actually fit between the elements indicated. Metals lie below and to the left of the thick "staircase" line [top of 3A(13) to bottom of 6A(16) in Period 6] and include main-group metals (purple-blue), transition elements (blue), and inner transition elements (gray-blue). Nonmetals (yellow) lie to the right of the line. Metalloids (green) lie along the line. We discuss the placement of hydrogen in Chapter 14. As of mid-2007, elements 112–116 had not been named.

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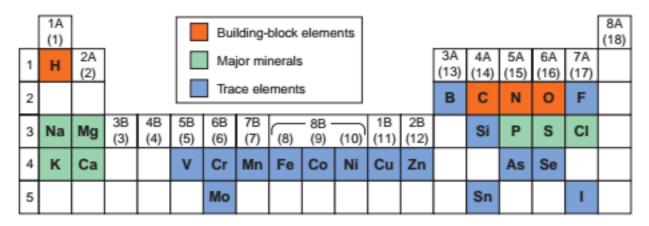
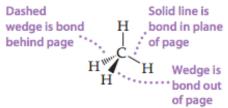


Figure 2.17 A biological periodic table. The building-block elements and major minerals are required by all organisms. Most organisms, including humans, require the trace elements as well. Many other elements (not shown) are found in organisms but have no known role.

CH_4

Molecular formula

Structural formula



Perspective drawing



Ball-and-stick model



Space-filling model