Answers to problems in pages 567-568

Practice Problems

- **40.** 6.13×10^{-7} M
- **41.** 2.74×10^{-14} M
- **42.** 1.7×10^{-3} mol
- **43.** 5.35×10^{-12} M
- **44.** $[OH^-] = 0.300 \text{ M};$ $[H_3O^+] = 3.33 \times 10^{-14} \text{ M}$
- **45.** 1.41×10^{-7} M
- 46. 1.52
- **47.** 12.13
- **48**. 0.82
- 49. 12.91
- **50.** 0.54
- 51. a. 2.3
 - **b.** 1.3
 - c. 0.3
 - **d.** -0.7
- **52.** 13.81
- **53**. 13.48
- **54.** 9.00
- **55.** 0.17