## Answers to problems in page 228.

Answers to Practice
Problems A

1. $1.13 \times 10^{23}$ ions $\mathrm{Na}^{+}$
2. $8.73 \times 10^{6}$ atoms As
3. $2.544 \times 10^{24}$ molecules $\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{O}_{2}$
4. $3.6 \times 10^{24}$ formula units NaOH

Homework - Genirme
Additional Practice How many representative particles are present in each of the following?

1. 4.3 mol of tungsten Ans. $2.6 \times$ $10^{24}$ atoms W
2. $2.45 \times 10^{-6} \mathrm{~mol}$ of nickel(II) selenide Ans. $1.48 \times 10^{18}$ formula units NiSe
3. 0.923 mol of selenium tetrabromide Ans. $5.56 \times 10^{23}$ molecules $\mathrm{SeBr}_{4}$
LS Logical
