## **Answers to Practice Problems in page 243**

## Answers to Practice Problems G 1. Mn<sub>2</sub>O<sub>3</sub>

- **2.** Cl<sub>2</sub>O<sub>7</sub>
- 3. Fe<sub>3</sub>O<sub>4</sub>
- 4. C<sub>2</sub>H<sub>3</sub>Cl<sub>3</sub>



Additional Practice For each of the following examples, find the empirical formula for a compound with the given composition.

- 1. 63.52% Fe, 36.48% S Ans. FeS
- 2. 26.58% K, 35.35% Cr, 38.07% O Ans. K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>
- 32.37% Na, 22.58% S, 45.05% O Ans. Na<sub>2</sub>SO<sub>4</sub>
- **4.** 74.51% Pb, 25.49% Cl Ans. PbCl<sub>2</sub>
- 5. 52.55% Ba, 10.72% N, 36.73% O Ans. BaN<sub>2</sub>O<sub>6</sub> or Ba(NO<sub>3</sub>)<sub>2</sub>
- 6. 29.15% N, 8.407% H, 12.50% C, 49.94% O Ans. N<sub>2</sub>H<sub>8</sub>CO<sub>3</sub> or (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> └ Logical