

Answers to Practice Problems F in page 239

Answers to Practice Problems F

- 259.80 g/mol
 - 136.06 g/mol
 - 342.34 g/mol
 - 253.80 g/mol
 - 60.06 g/mol
 - 262.84 g/mol
- NaHCO_3 , 84.01 g/mol
 - CeB_6 , 204.98 g/mol
 - $\text{Mg}(\text{ClO}_4)_2$, 223.20 g/mol
 - $\text{Al}_2(\text{SO}_4)_3$, 342.17 g/mol
 - $\text{Fe}(\text{OH})_3$, 106.88 g/mol
 - SnCl_2 , 189.61 g/mol
 - P_4O_{10} , 283.88 g/mol
 - ICl , 162.35 g/mol
- 92.15 g/mol
 - 0.0815 mol $\text{C}_6\text{H}_5\text{CH}_3$
- 300.06 g/mol
 - 2.050 g $\text{PtCl}_2(\text{NH}_3)_2$

Homework — GENERAL

Additional Practice Calculate the molar mass of each of the following compounds:

- KNO_3 Ans. 101.11 g/mol
- Na_2SO_4 Ans. 142.05 g/mol
- $\text{Ca}(\text{OH})_2$ Ans. 74.10 g/mol
- $(\text{NH}_4)_2\text{SO}_3$ Ans. 116.17 g/mol
- $\text{Ca}_3(\text{PO}_4)_2$ Ans. 310.18 g/mol
- $\text{Al}_2(\text{CrO}_4)_3$ Ans. 401.96 g/mol

 Logical