## Answers to Practice Problems E in page 435

## Answers to Practice

Problems E

1. $7.97 \times 10^{-2} \mathrm{~mol}$
2. 0.137 mol
3. 1500 kPa
4. $2.73 \times 10^{4} \mathrm{~L}$

Homework - General

## Additional Practice

1. How many moles of gas are in a balloon that has a volume of 15.9 L at a pressure of 149 kPa and a temperature of $28^{\circ} \mathrm{C}$ ? Ans. 0.947 mol
2. How many moles of ammonia, $\mathrm{NH}_{3}$, is required to fill a 14.88
L bottle to a pressure of 199 kPa at $25^{\circ} \mathrm{C}$ ? Ans. $1.20 \mathrm{~mol} \mathrm{NH}_{3}$
3. What is the volume of 4.35 moles of a gas at a pressure of 85.6 kPa and $26.0^{\circ} \mathrm{C}$ ? Ans. 126 L
